



Erratum

Erratum to “Synthesis and enantiomeric resolution of ferrocenyl(alkyl)azoles” [Journal of Organometallic Chemistry 688 (2003) 138–143] ^{☆,☆☆}

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The publisher regrets that there were a number of errors in the reference list of the above-mentioned paper. The correct list is given below.

References

- [1] T.S. Lobana, in: F.R. Hartley (Ed.), *The Chemistry of Organophosphorus Compounds*, vol. 2, Wiley, New York, 1992 (Chapter 8).
- [2] A.K. Bhattacharya, N.K. Roy, in: F.R. Hartley (Ed.), *The Chemistry of Organophosphorus Compounds*, vol. 2, Wiley, New York, 1992 (Chapter 6).
- [3] J. Halpern, B.L. Goodall, G.P. Khare, H.S. Lim, J.J. Plath, *J. Amer. Chem. Soc.* 97 (1975) 2301.
- [4] J. Halpern, A.L. Pickard, *Inorg. Chem.* 9 (1970) 2798.
- [5] A.R.J. Genge, W. Levenson, G. Reid, *Inorg. Chim. Acta* 288 (1999) 142.
- [6] W. Du Mont, B. Neudert, H. Schumann, *Angew. Chem. Int. Ed. Eng.* 15 (1976) 308;
W. Du Mont, H.-J. Kroth, H. Schumann, *Chem. Ber.* 109 (1976) 3017.
- [7] C. Glidewell, *J. Organomet. Chem.* 116 (1976) 199.
- [8] The use of hydrogen peroxide can be hazardous if some of the H₂O₂ remains in the product. On one occasion a sample of solid *o*-C₆H₄(P(O)Ph₂)₂ detonated with great violence whilst being dried on a vacuum line, almost certainly because it retained hydrogen peroxide of crystallisation, i.e., the hydrogen peroxide co-crystallised with the phosphine oxide.
- [9] A.R.J. Genge, Ph.D. Thesis, Southampton University, 1998.
- [10] A.M. Aguiar, J. Beisler, *J. Org. Chem.* 29 (1964) 1660.
- [11] H.C.E. McFarlane, W. McFarlane, *Polyhedron* 2 (1983) 303.

* PII of original article S0022-328X(03)00915-X.

☆☆ DOI of original article 10.1016/j.jorganchem.2003.11.008.

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